

UNCLASSIFIED

Alteration of type-G wellbore cement during carbon sequestration: An experimental case study

Marcus Wigand, J. William Carey, W. Kirk Hollis, John P. Kaszuba

Los Alamos National Laboratory

LA-UR-071358



The World's Greatest Science Protecting America

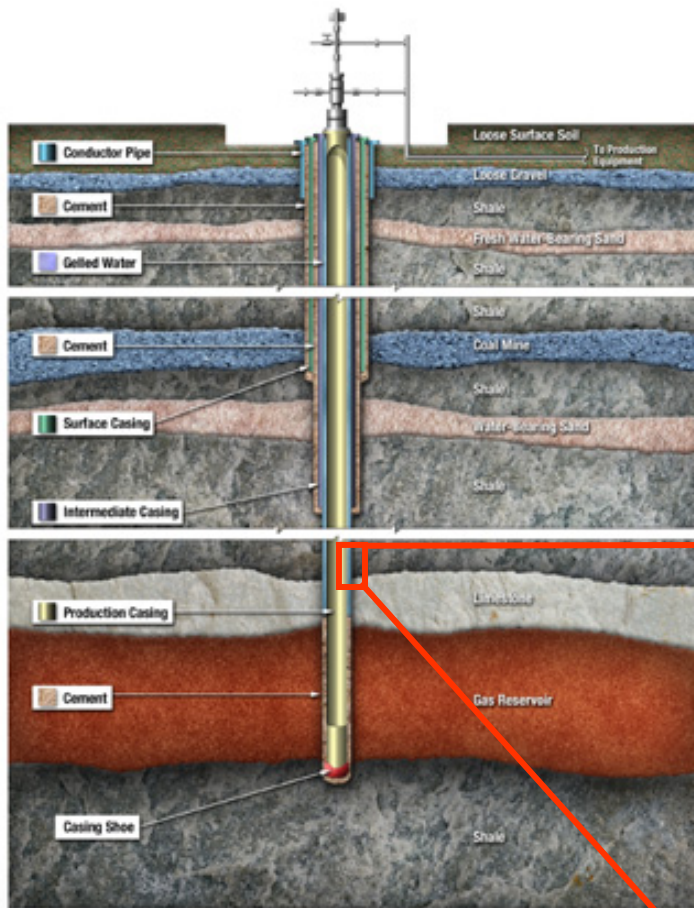
UNCLASSIFIED



Objectives for this study

- Experimental study of the wellbore class-G cement-formation water-SCCO₂ system under in-situ reservoir conditions during carbon sequestration
- How does reaction with SCCO₂ affect the geophysical properties (e.g. porosity and permeability) and integrity of the cement?
- Can Alkali-Carbonate Reactivity (ACR) processes occur in wellbore cement during carbon sequestration?

Wellbore integrity

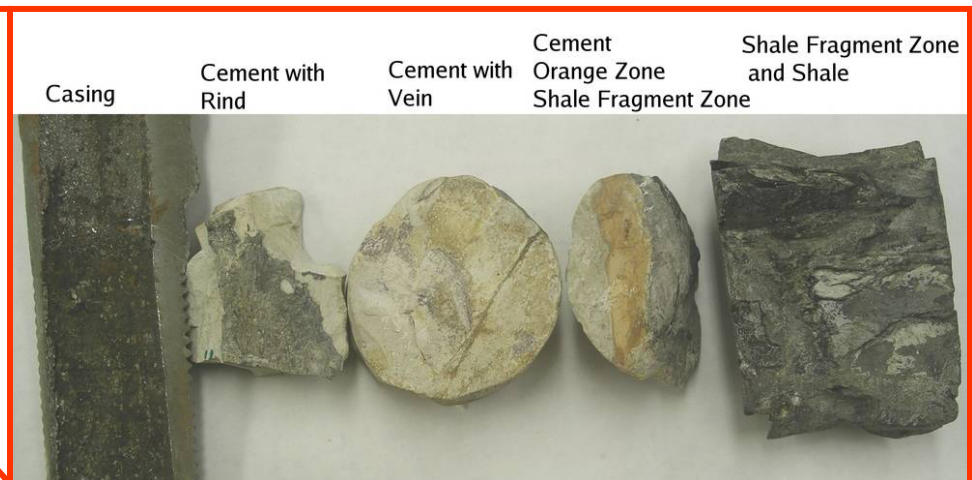


Casing Design in a Typical Gas Well
(Not to Scale)

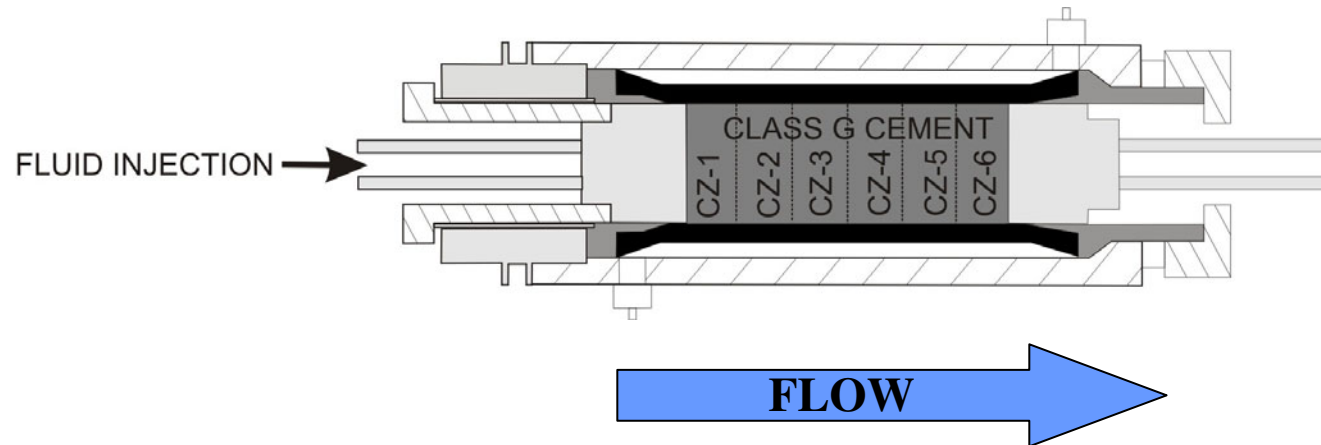
After www.envisioncreativegroup.com

Potential risks:

- ❖ Wellbore flow
- ❖ Cement interactions with supercritical CO₂ and/or acid gases
- ❖ Cement interaction with formation brine



Experimental Setup



EXPERIMENTAL SETUP:

Hassler vessel

Injection pressure 2880 psi

Pressure gradient 40 psi

Confining pressure 3800 psi

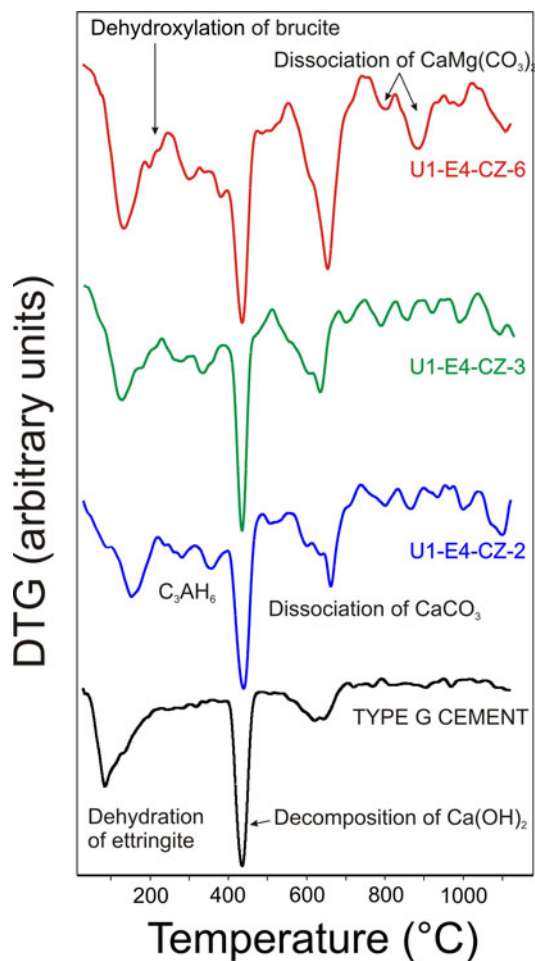
Temperature 54°C

PROCEDURE:

Saturation with 1.65 M brine over 31 days

Injection of SCCO₂ over 141 days

Mineralogy

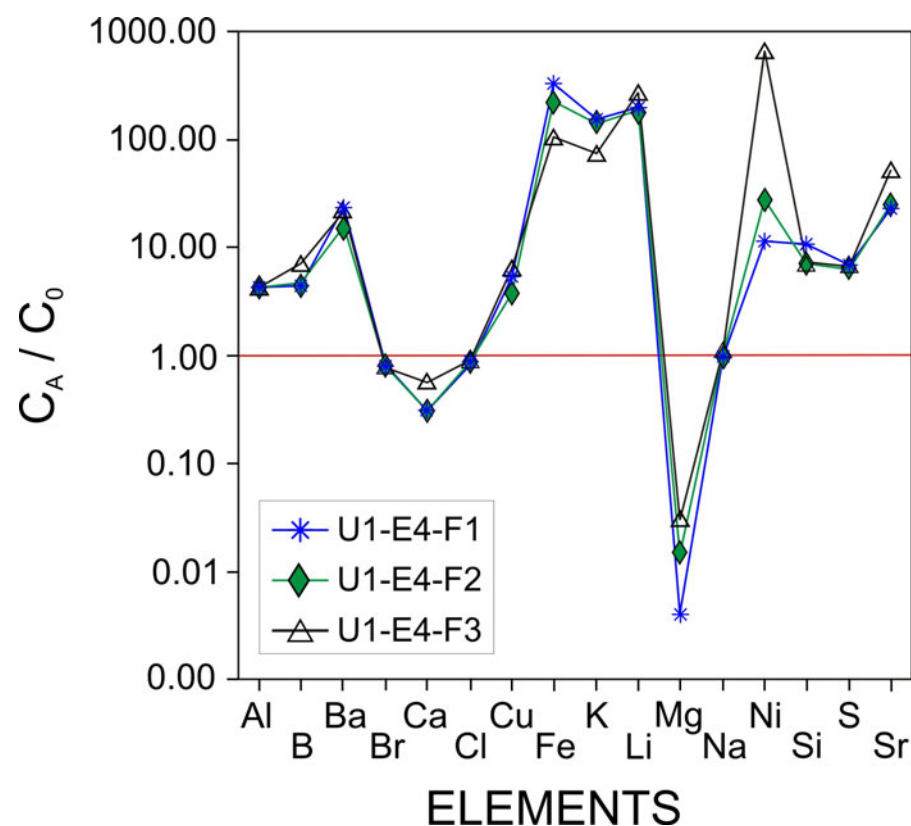


	CLASS-G CEMENT	U1-E4-CZ1	U1-E4-CZ2 TO U1-E4-CZ5	U1-E4-CZ6
$\text{Ca}_5\text{Al}_2(\text{SiO}_4)_3(\text{OH})_4$	X	X	X	X
$\text{Ca}(\text{OH})_2$	X	X	X	X
$\text{Ca}_3\text{Al}_2(\text{OH})_{12}$	X	X	X	X
$\text{Ca}_{1.5}\text{SiO}_{3.5} \cdot \text{H}_2\text{O}$	X	X	X	X
$\text{Ca}_2(\text{Al,Fe})_2\text{O}_5$	X	X	X	X
Ca_3SiO_5	X			
$\text{Al}(\text{OH})_3$	X			
$\text{Ca}_4\text{Al}_2\text{O}_6\text{Cl}_2 \cdot 10 \text{H}_2\text{O}$		X	X	X
$\text{Mg}(\text{OH})_2$				X
Calcite	X	X	X	X
Aragonite				X
$\text{Ca Mg}(\text{CO}_3)_2$				X

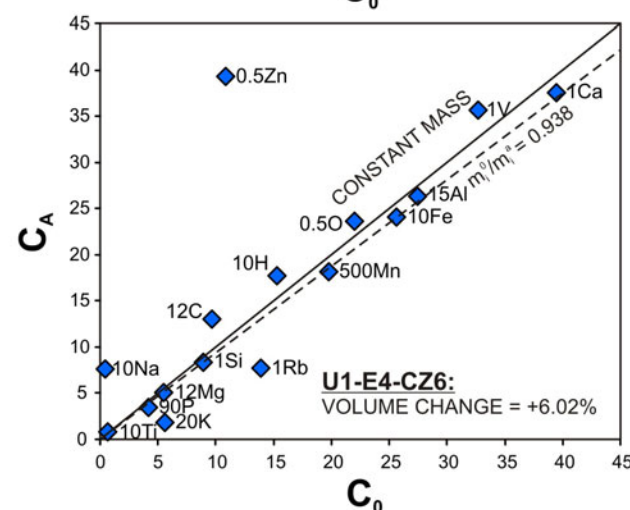
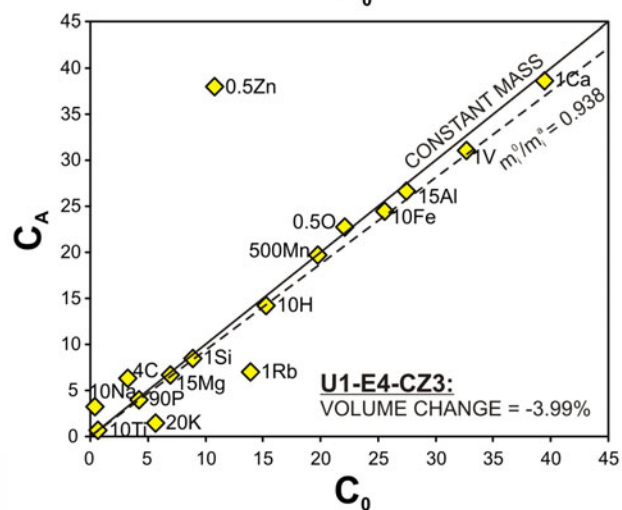
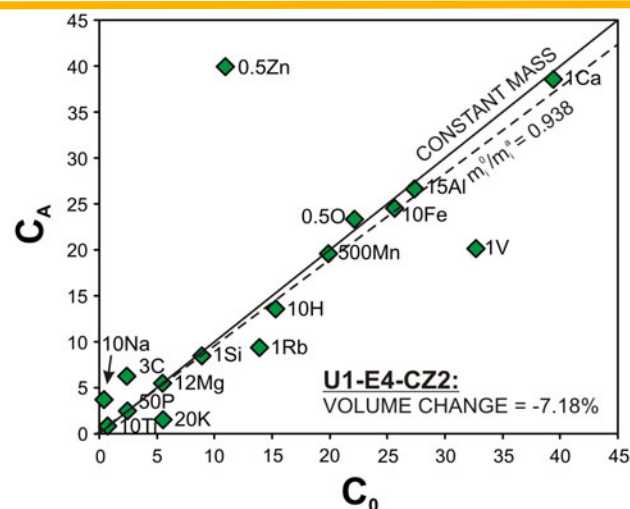
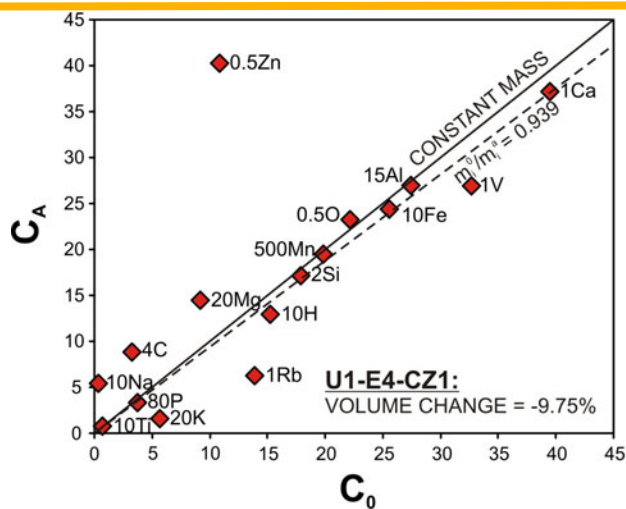
Fluid Chemistry

Brine Composition:

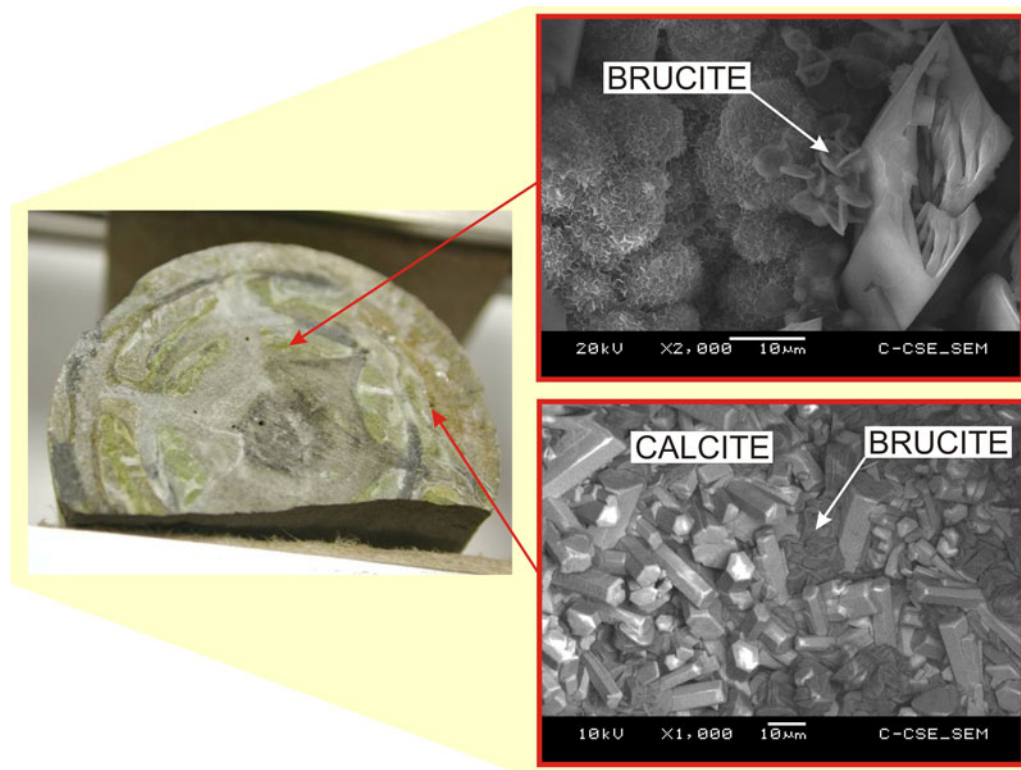
pH	6.61
HCO_3^-	1181 ppm
Cl^-	53493 ppm
SO_4^{2-}	507 ppm
Na^+	32525 ppm
Ca^{2+}	4284 ppm
Mg^{2+}	724 ppm
Ionic strength	1.651
TDS	88953 mg/kg



Whole-Rock Geochemistry



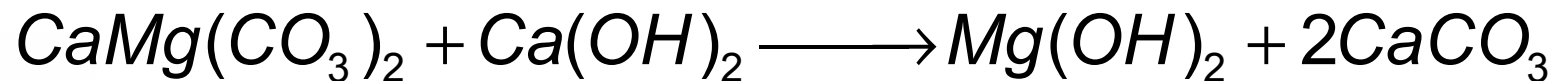
Alkali-Carbonate Reactivity Products at the opposite side of the core



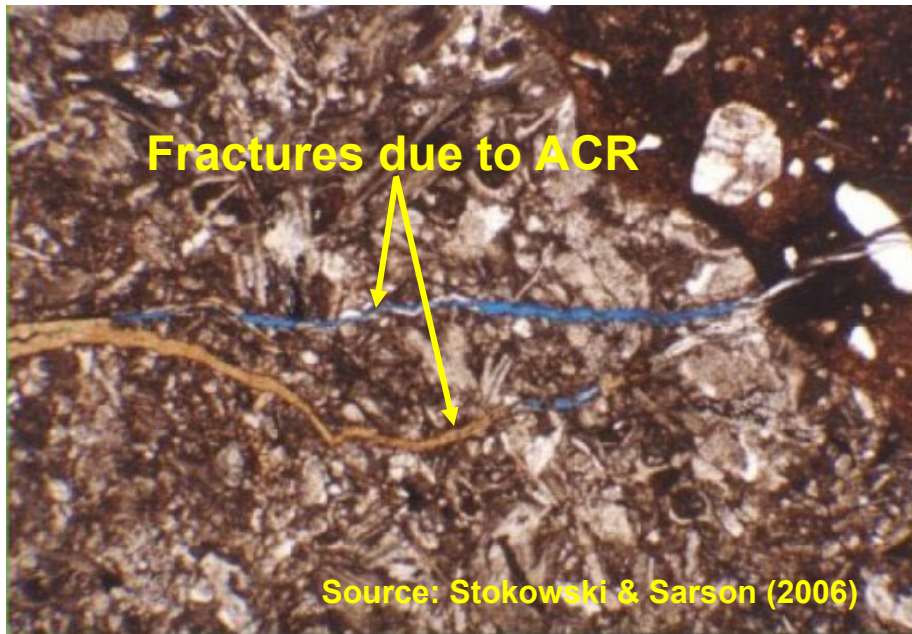
XRD RESULTS:

CEMENT PHASES	wt%
Portlandite	15
Brownmillerite	9
Hydrocalumite	7
Katoite	6
Ca ₂ SiO ₄	7
Ca ₃ SiO ₅	4
Calcite	6
Aragonite	2
Brucite	4
Amorphous	40

ACR-TYPE REACTION (“DEDOLOMITIZATION”):



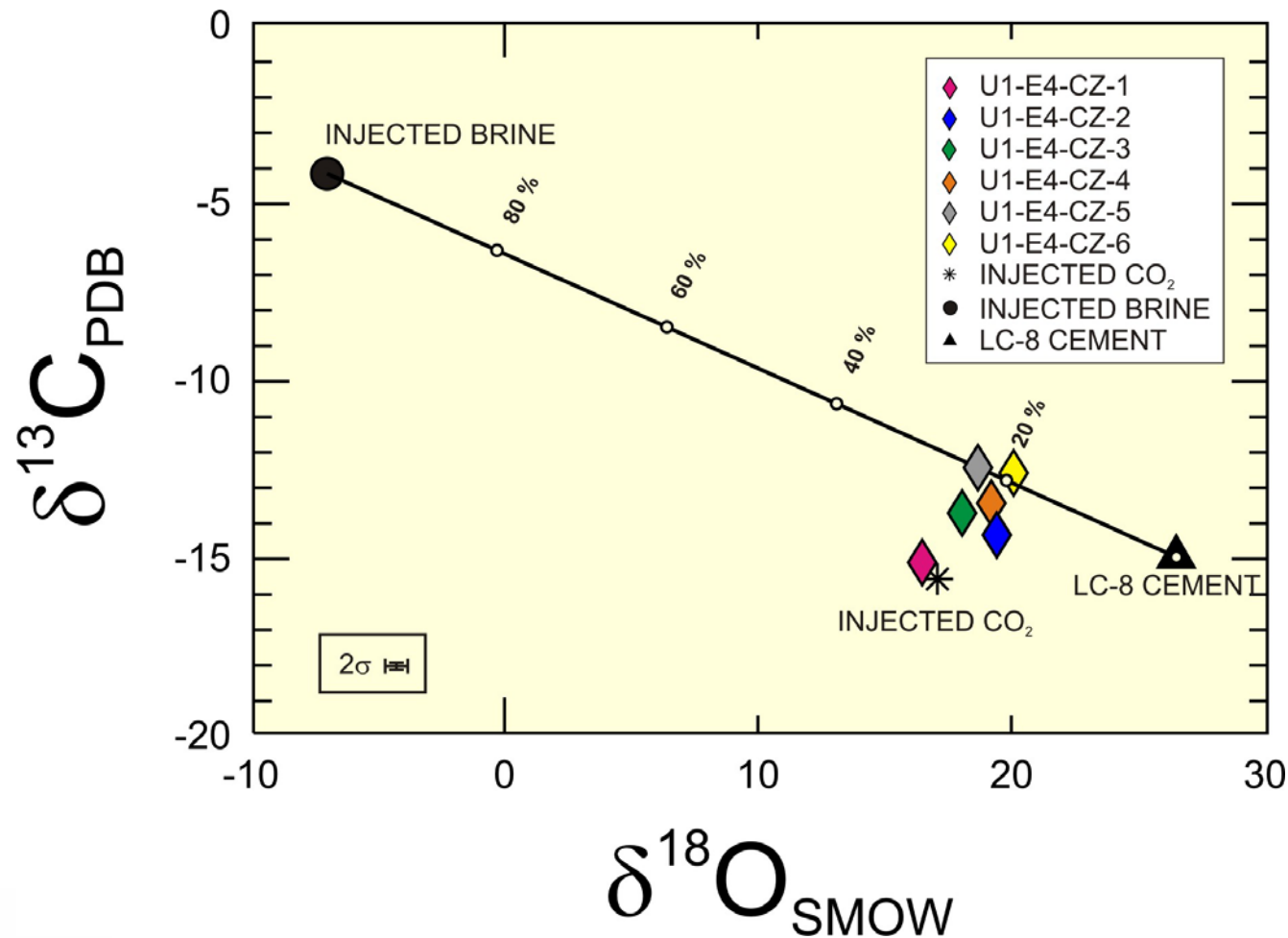
Alkali-Carbonate Reactivity



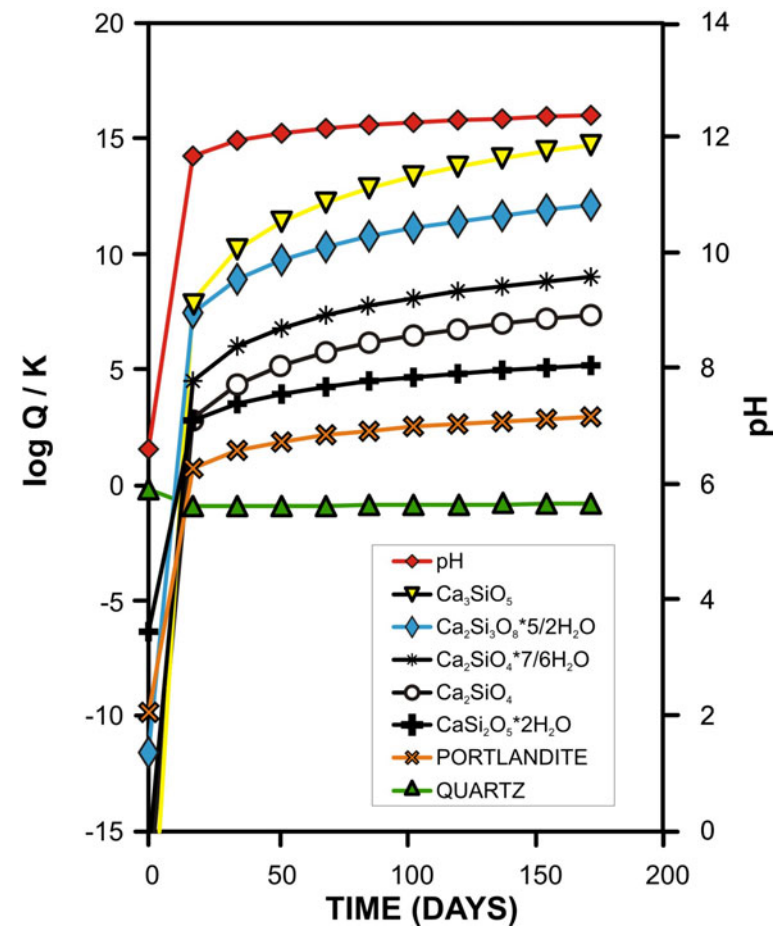
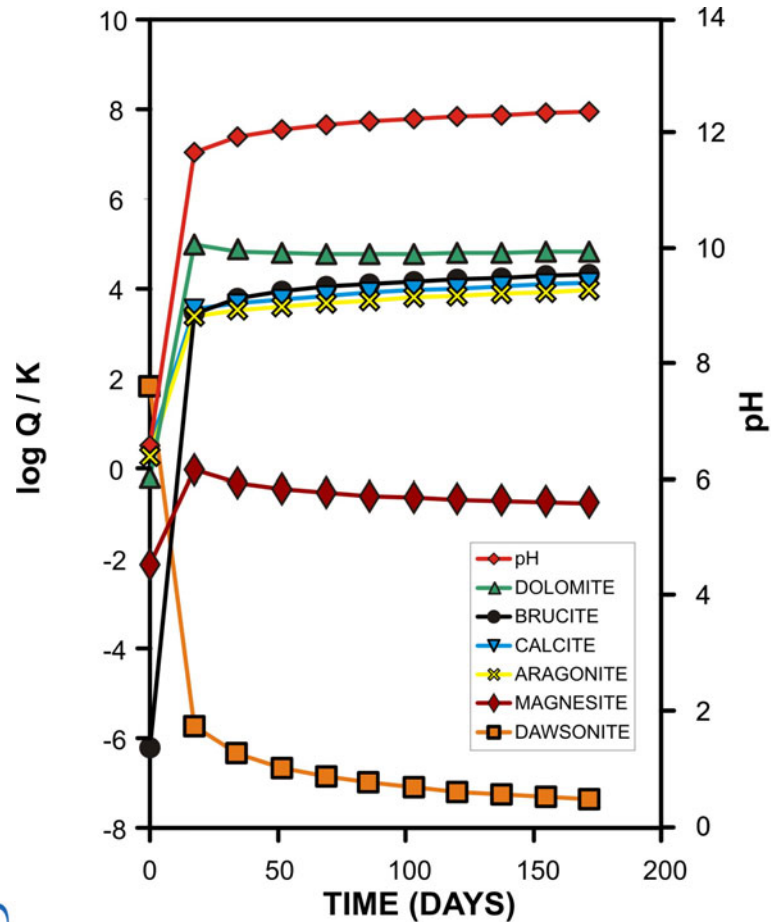
Results of ACR:

- Expansion of structural members due to differential volume change
- Formation of fractures and loss of bond between aggregate particles and cement paste
- Brucite and Ca-rich deposits in cracks, voids, and paste

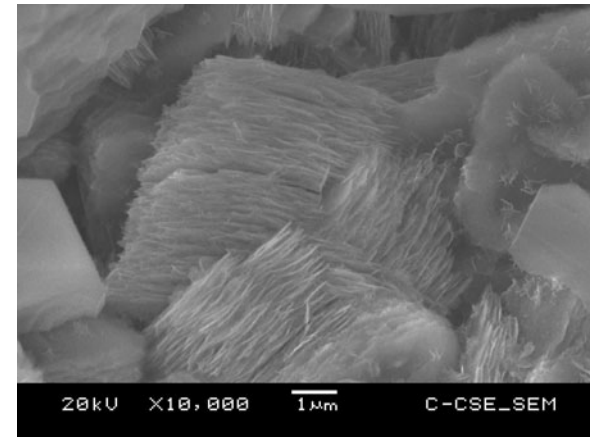
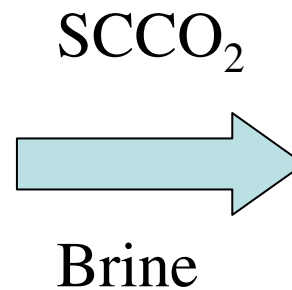
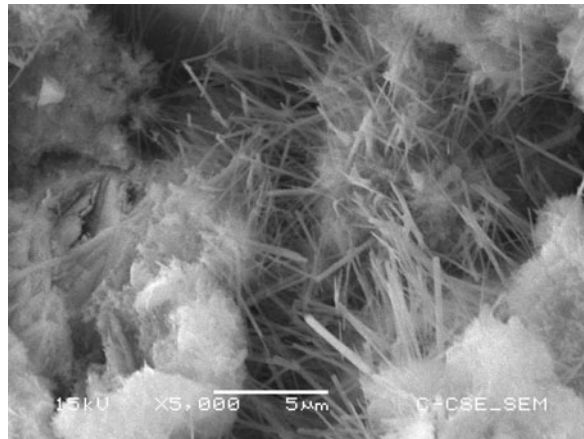
Stable Isotopes as an Indicator for the Carbonation Process of the Wellbore Cement



Geochemical Modeling



Changes in Cement Texture



- ❖ Changes in the texture of the cement but no formation of an orange-colored carbonation zone
- ❖ Increase in weight by 7.13% and decrease in porosity by 11.2%



Conclusions

- ❖ Not every Portland-based wellbore cement shows a complete carbonation during the reaction with SCCO_2 & brine
- ❖ Portlandite and C-S-H present throughout the core
- ❖ No formation of an orange alteration zone
- ❖ CO_2 does not appear to have penetrated through the length of the core (2.39 cm) during the 172 day experiment
- ❖ ACR-like processes can occur during carbon sequestration processes

Acknowledgements

SPECIAL THANKS TO:

Toti Larson

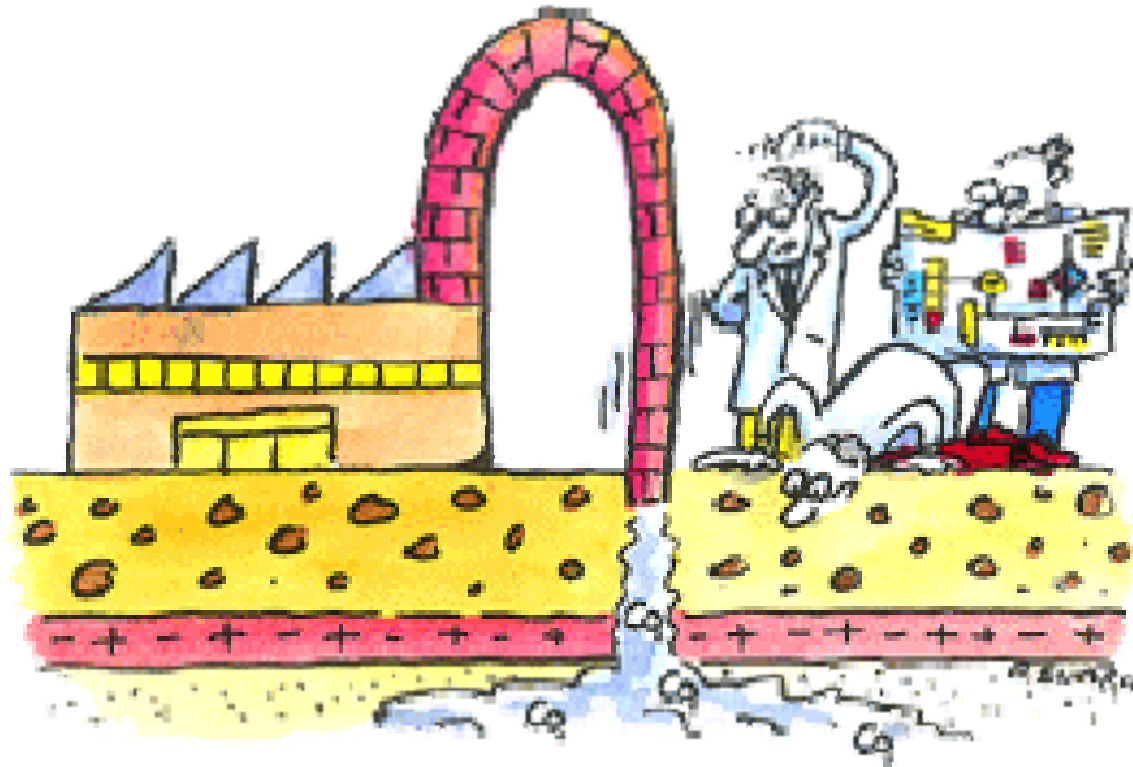
Melissa Fittipaldo

Dale Counce

Ren-Guan Duan

Steve Chipera

Emily Kluk



Research funded by Los Alamos National Laboratory and by the US Department of Energy

UNCLASSIFIED

Thank you very much for your attention !



UNCLASSIFIED